

FIITJEE INTERNAL TEST

FIRST YEAR 2020-22

JEE MAINS 19th JULY 2020

ANSWER & KEY

MATHEMATICS (Finalised by MKK)

1. B	2. D	3. B	4. B	5. B
6. B	7. D	8. D	9. D	10. C
11. D	12. B	13. B	14. D	15. B
16. C	17. A	18. C	19. B	20. C
21. 4	22. 56	23. 3	24. 3	25. 8

PHYSICS (Finalised by PRKH)

26. C	27. D	28. A	29. C	30. B
31. A	32. C	33. A	34. BONUS	35. D
36. C	37. C	38. C	39. B	40. C
41. A	42. B	43. D	44. A	45. A
46. 4	47. 1	48. 9	49. 4	50. 5

CHEMISTRY (Finalised by CCP)

51. D	52. B	53. A	54. C	55. C
56. B	57. C	58. A	59. D	60. C
61. B	62. C	63. C	64. A	65. D
66. B	67. C	68. B	69. BONUS	70. C
71. 5	72. 2	73. 4	74. 2	75. 2

CHEMISTRY SOLUTIONS

$$51. \text{P.E.} = \frac{-kze^2}{r} = \frac{-3e^2}{4\pi\epsilon_0 r}$$

$$v_1 = x$$

$$52. v_3 = x \left(\frac{1}{3} \right) = \frac{x}{3}$$

$$53. mvr = \frac{nh}{2\pi}$$

$$= \frac{5h}{2\pi} = \frac{2.5h}{\pi}$$

$$54. \frac{\text{KE}}{\text{TE}} = \frac{\left(\frac{k \cdot ze^2}{2r} \right)}{\left(\frac{-kze^2}{2r} \right)} = -1$$

$$55. r_{2(\text{Be}^{3+})} = 0.529 \times \frac{4}{4} = 0.529 = r_{1(\text{H})}$$

$$56. E_n = -13.6 \times \left(\frac{Z^2}{n^2} \right) \text{eV/atom}$$

$$\therefore \frac{(E_2 - E_1)}{(E_4 - E_3)} = 15.4$$

$$57. T \propto \left(\frac{n^3}{Z^2} \right) \Rightarrow T \propto n^3$$

$$\frac{T_1}{T_2} = \frac{n_1^3}{n_2^3}$$

$$59. \frac{1}{\lambda} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2}$$

$$60. E = -13.6 \times 4 = -54.4 \text{eV}$$

$$\therefore \text{IE} = 54.4 \text{eV}$$

$$61. \Delta x = \frac{h}{4\pi \cdot m \Delta V}$$

$$62. r = \frac{n^2}{Z} \left[\frac{h^2}{4\pi^2 m k e^2} \right] = \frac{h^2 a_0}{Z}$$

$$\therefore v_n = \frac{hz}{2\pi a_0 n}$$

$$\therefore \text{KE} = \frac{1}{2} m v^2 = \frac{h^2 z^2}{8\pi^2 m a_0^2 n^2}$$

$$\therefore \text{KE} = \frac{h^2}{32\pi^2 m a_0^2}$$

$$66. \frac{1}{\lambda} = \frac{1}{\lambda_1} + \frac{1}{\lambda_2} \Rightarrow \frac{1}{\lambda} = \frac{\lambda_1 + \lambda_2}{\lambda_1 \lambda_2}$$

$$\Rightarrow \lambda = \frac{\lambda_1 \lambda_2}{\lambda_1 + \lambda_2}$$

$$67. hv = hv_0 + \text{K.E}$$

$$\frac{hc}{\lambda} = \frac{hc}{\lambda_0} + \frac{1}{2}mv^2 \Rightarrow \frac{1}{2}mv^2 = hc \left[\frac{1}{\lambda} - \frac{1}{\lambda_0} \right]$$

$$V = \sqrt{\frac{2hc}{m} \left[\frac{\lambda_0 - \lambda}{\lambda\lambda_0} \right]}$$

$$68. \quad mvr = \frac{nh}{2\pi} = 3.1652 \times 10^{-34} \Rightarrow n = 3$$

$$n_2 = 3, n_1 = 2; \frac{1}{\lambda} = 4R_H \left[\frac{1}{4} - \frac{1}{9} \right] \Rightarrow \frac{1}{\lambda} = 4R \frac{5}{36} = \frac{5R}{9}$$